## Dispersant from Deepwater Horizon spill found to persist in the environment

July 16 2014



Helen White on a Gulf coast beach holding a jetty rock with oil residue on it. Credit: Mike McNulty, Woods Hole Oceanographic Institution

(Phys.org) —The 2010 Deepwater Horizon (DWH) spill in the Gulf of Mexico was the largest accidental release of oil into the ocean, with approximately 210 million gallons gushing from the blown out well. In

an attempt to prevent vast quantities of oil from fouling beaches and marshes, BP applied 1.84 million gallons of chemical dispersant to oil to oil released in the subsurface and to oil slicks at the sea surface. The dispersant was thought to rapidly degrade in the environment.

A new study by scientists at Haverford College and Woods Hole Oceanographic Institution (WHOI) has found that the <u>dispersant</u> compound DOSS, which decreases the size of <u>oil droplets</u> and hampers the formation of large oil slicks, remains associated with oil and can persist in the environment for up to four years.

The study was recently published in *Environmental Science & Technology Letters*.

The study examined samples collected from deep-sea corals and surrounding sediments collected in Dec. 2010 as well as oil-soaked sand patties found on coastal beaches since July 2010 to the present.

"We found DOSS persisted in variable quantities in deep-sea coral communities 6 months after the spill and on Gulf of Mexico beaches 26–45 months after the spill," said lead author Helen White, an assistant professor of chemistry with Haverford College. "These results indicate that the dispersant, which was thought to undergo rapid degradation in the water column, remains associated with oil in the environment and can persist for around 4 years."

Finding dispersant in the deep-sea samples they analyzed did not come as a surprise to the researchers. Earlier work by co-author Elizabeth Kujawinski and her colleagues found DOSS was slow to degrade in the deep ocean. "The deep sea is cold and dark and the degradation of dispersant components happens much more slowly under these conditions," said Kujawinski, an associate scientist at WHOI. "The interesting thing is that the sand patties we're finding on beaches four

years after the spill have DOSS in them. That was somewhat unexpected."



A helper from Chris Reddy's WHOI lab collected oil samples from a Gulf coast beach. Credit: Chris Reddy, Woods Hole Oceanographic Institution

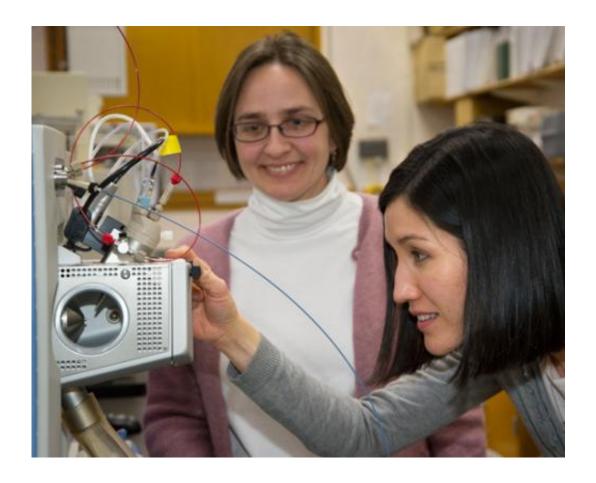
White and her colleague Chris Reddy at WHOI have been monitoring the beaches for oiled samples since the time of the spill. This fieldwork is part of an effort to gain as much understanding about the geographical and temporal distribution of the sand patties.

For this study, which was funded by the Gulf of Mexico Research Initiative (GOMRI), White and her team at Haverford developed a method to isolate the DOSS from the solid sand patties. Prior to this study, dispersant had only been analyzed in aqueous samples. White then sent the isolated compounds to Kujawinski's lab at WHOI, where researchers used sophisticated instruments to quantify the DOSS in samples collected from environments known to contain oil persisting from the DWH oil spill.

"The amounts we detected are quite small, but we're finding this compound in locations where we expected the dispersants to disappear, either by dissolving in water or by being degraded by bacteria," said Kujawinski.

What does the presence of DOSS in deep-sea sediments and corals and in sand patties on Gulf beaches mean for marine life or for the people who frequent the beaches? The scientists say the honest answer is they don't really know.

"It's hard to say because we don't know how toxic it might be. The EPA has determined what concentrations of DOSS may be harmful to marine life in the water, but the toxicity of DOSS in solid (non-aqueous) forms like sediments or sand patties is not known," said White. "We know that if you measure 'x' amount of this compound in 'y' amount of water, that's toxic. But you can't compare those numbers to what we've found in the sand patties because we're looking at this compound in a mixture of sand and oil."



Elizabeth Kujawinski, left, and her colleague Melissa Soule monitor a mass spectrometer. Credit: Tom Kleindinst, Woods Hole Oceanographic Institution

Until more is known about the toxicity of oil and dispersant mixtures, some Gulf beaches have signs to warn people not to touch the sand patties.

The researchers are eager for their findings to be known by other scientists studying the Deepwater Horizon spill. "We want to get this information out there because there are other scientists who are studying similar samples," said White. "When they ask questions about which microbes are living in the samples or how the oil is changing over time, it's really important to consider that the samples also contain dispersants."

The researchers hope to focus future research on gaining a better understanding of why dispersant compounds are able to persist in the environment for so long. In particular, they are interested in learning how they escape degradation in coastal environments where temperatures are higher for a significant part of the year and there is more exposure to sunlight.

"The dispersant was not expected to persist for so long," Kujawinski said. "Now that we know that it can be found up to 4 years after the spill, there is even more need to understand it and any impacts it may have on the environment."

## Provided by Woods Hole Oceanographic Institution

Citation: Dispersant from Deepwater Horizon spill found to persist in the environment (2014, July 16) retrieved 3 October 2025 from <a href="https://phys.org/news/2014-07-dispersant-deepwater-horizon-persist-environment.html">https://phys.org/news/2014-07-dispersant-deepwater-horizon-persist-environment.html</a>

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