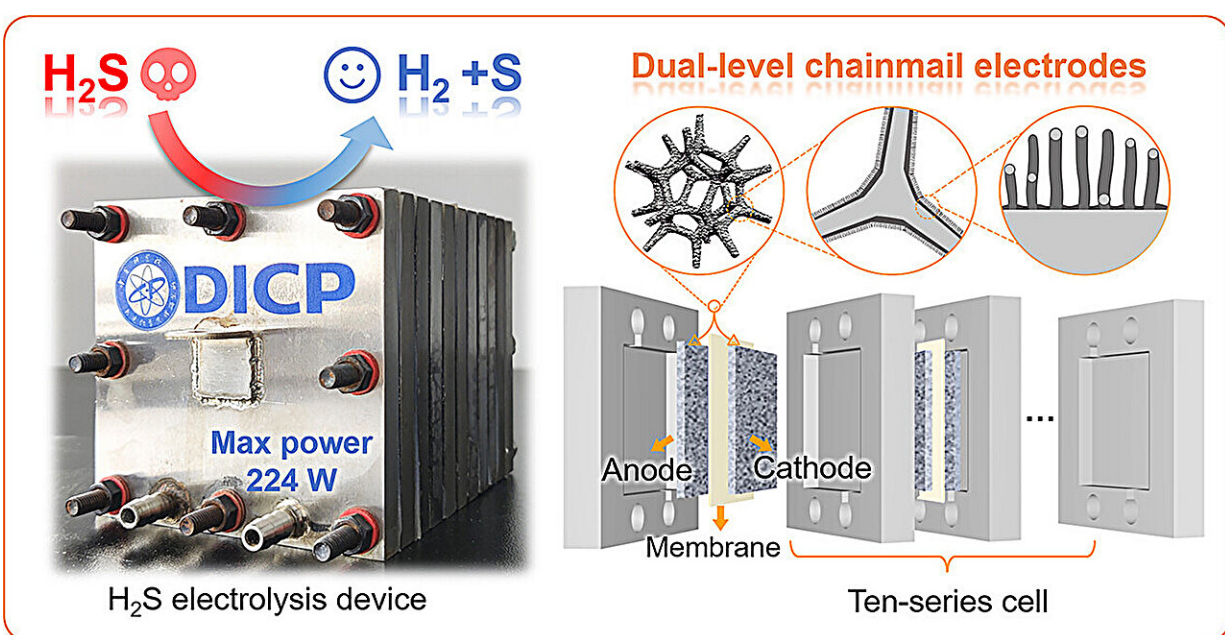


# Researchers develop chainmail integrated-electrode for highly efficient hydrogen sulfide electrolysis

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Graphical abstract. Credit: *Angewandte Chemie International Edition* (2025).  
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Hydrogen sulfide (H<sub>2</sub>S), a toxic and corrosive byproduct of fossil fuel extraction, poses significant environmental and industrial challenges. While the conventional Claus process converts H<sub>2</sub>S into elemental sulfur, it fails to recover hydrogen gas, missing an opportunity for sustainable

energy production.

Electrocatalytic H<sub>2</sub>S decomposition offers a promising strategy to simultaneously eliminating pollutants and producing green hydrogen. However, the acidic nature of H<sub>2</sub>S deactivates non-precious metal catalysts and degrades electrode structures, making it difficult to achieve both [high efficiency](#) and long-term stability.

In a study published in [Angewandte Chemie International Edition](#), a research group led by Prof. Deng Dehui and Assoc. Prof. Cui Xiaojun from the Dalian Institute of Chemical Physics (DICP) of the Chinese Academy of Sciences has developed a dual-level chainmail integrated-electrode that enables highly efficient hydrogen production via H<sub>2</sub>S electrolysis.

Researchers designed a graphene encapsulating nickel foam (Ni@NC foam) electrode with a dual-level chainmail structure, enhancing both catalytic activity and durability.

This electrode achieved an industrial-scale current density exceeding 1 A/cm<sup>2</sup> at 1.12 V versus the reversible hydrogen electrode, which is five times higher than commercial nickel foam electrodes. The Ni@NC foam electrode remained stable for over 300 hours, demonstrating a lifespan at least ten times longer than commercial nickel foam electrodes.

In a simulated natural gas desulfurization test, the chainmail integrated-electrode completely oxidized and removed 20% H<sub>2</sub>S at the anode, producing sulfur powder simultaneously. Meanwhile, high-purity hydrogen was collected at the cathode. Compared with conventional water electrolysis, the system reduced [energy consumption](#) by 43% at the [current density](#) of 200 mA/cm<sup>2</sup>, offering a more sustainable approach to hydrogen production.

"Our study provides an efficient, low-energy solution for natural gas purification and opens up the potential of converting H<sub>2</sub>S into valuable hydrogen fuel for industrial applications," said Prof. Deng.

**More information:** Mo Zhang et al, Highly Effective and Durable Integrated-Chainmail Electrode for H<sub>2</sub> Production through H<sub>2</sub>S Electrolysis, *Angewandte Chemie International Edition* (2025). DOI: [10.1002/anie.202502032](https://doi.org/10.1002/anie.202502032)

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